## SDS gels to identify HBB gene products in the ts *fla* mutants



Aizawa et al., J. Bacteriol. (1985)

## Ring preparations from HBB



Appare mol wt (	ent (10 <sup>3</sup> ) pI	Gene	Morphological features, functions or comments
65	6.8–7.3	flaAII.1 <sup>b</sup>	M ring? Base plate for energy-transducing and switching proteins?
60	5.0	flaW <sup>c</sup>	Junction between hook and filament; absent in polyhook-basal body complexes; found as HAP1 in hooks from filamentless mutants
58	4.8	$H2^d$	Flagellar filament, 1,2 antigen
53	5.2	$Hl^d$	Flagellar filament, i antigen
42	4.7	flaFV <sup>e</sup>	Hook protein
38	ca.9	flaFIX <sup>b</sup>	P ring portion of outer cylinder?
32	4.5	flaFVI <sup>b</sup>	Rod?
30	4.6	flaFVII <sup>b</sup>	Rod?
27	7.5	flaFVIII?	L ring and wall portion of outer cylinder?
16	5.4	?	?
14	5.5	?	?
12	7.3	?	Ŷ
	2	9	S ring



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## 蛍光付加物の作成



The amino acid sequence of a polypeptide chain.

Figure 7-7

 Figure 7-8a
 The generation of the gas phase ions required for the mass spectrometric analysis of proteins. (a) By electrospray ionization (ESI).

 LysやArgにプロトン化した電荷により分離 (0.5~2電荷/kD)



Figure 7-8b The generation of the gas phase ions required for the mass spectrometric analysis of proteins. (b) By matrix-assisted laser desorption/ionization (MALDI).



Figure 7-9

The ESI-MS spectrum of the 16,951-D horse heart protein apomyoglobin.







**Figure 7-8c** The generation of the gas phase ions required for the mass spectrometric analysis of proteins. (*c*) By fast atom bombardment (FAB).



Figure 7-10 The use of a tandem mass spectrometer (MS/MS) in amino acid sequencing.





(a) Electrospray ionization tandem mass spectrometer



Figure 7-11 The tandem mass spectrum of the doubly charged ion of the 14-residue human [Glu<sup>1</sup>]fibrinopeptide B (m/z = 786).



642.6 627.8

1.14

600

400

20

173.7

200

942.7

1000

800

m/z

1325.1 1396.8

1400

1200

Thermodiynamic Principles



熱力学の法則				化学ポテンシャル
<ol> <li>エンタルピーの定義:H=U+PV</li> <li>ΔH=ΔU+PΔV, ΔU=ΔQ-ΔW</li> </ol>	U:エネルギー P:圧力 V:体積 Q:熱			化学ポテンシャル 」(部分モル自由エネルギー)
(第一法則) ΔH = ΔQ - ΔW +PΔV = ΔQ - ΔW'	W : 仕事	out	in	$u_0 = u^{\circ} + RT \cdot lnA_0$
2) エントロピー:S dS = dQ/T (可逆過程)	蒸気になるときの エンタルピー変化	Ao	Ai	$\mu_{i} = \mu^{\circ} + RT \cdot InA_{i}$
本の蒸発のΔH <sub>vap</sub> =40.7 kJmol <sup>-1</sup> で T=373 °Kであるから 3) ギブスの自由エネルギー・C=H-TS	$\Delta S_{vap} = 109.1 \text{ JK}^{-1}$ 1N = 1 Kg · m · s <sup>-2</sup> [N · m] = [1]		Δμ = μ <sub>j</sub>	$-\mu_{O} = RT \cdot ln(A_{O}/A_{i})$
$\Delta G = \Delta H - T\Delta S ( % a & A + 1) $ $\Delta G = \Delta H - T\Delta S ( % a & A + 1) $ $1/2 \cdot mv^2 = 1/2 \cdot (2 \text{ kg}) \cdot (1 \text{ m m})$	$(1 + m_1 - 1)^2$ $(1 + m_1 - 1)^2 = 1$ Kg $\cdot$ m <sup>2</sup> $\cdot$ s <sup>-2</sup> $\cdot$ s <sup>-1</sup> ) <sup>2</sup> = 1 Kg $\cdot$ m <sup>2</sup> $\cdot$ s <sup>-2</sup> = 1Nm			T: Kelvin 温度 R: ガス定数(2cal・mol <sup>-1</sup> ・K <sup>-1</sup> )
質量2 kgが1 m・s <sup>-1</sup> の速 のの運動エネルギーに1	さで動いているも JJが対応			