問題

Acetaldehyde + NADH + $H^+ \longrightarrow ethanol + NAD^+$

- この反応を 2 つの半反応式に分けることが出来る (1) Acetaldehyde + $2H^+ + 2e^- \longrightarrow$ ethanol (2) NAD⁺ + $2H^+ + 2e^- \longrightarrow$ NADH + H⁺ $E'_0 = -0.320$ V
- 上の式で全ての物質が一モル存在したときの自由エネ ルギー変化をもとめよ。ファラデー常数は96.5 kJ/V・ molとせよ。
- 2) 膜蛋白質の性質について説明せよ。

答案用紙に名前を書くのを忘れないこと。















Figure 22-11 Effect of inhibitors on electron transport.



Figure 22-13 Determination of the stoichiometry of coupled oxidation and phosphorylation (the P/O ratio) with different electron donors.

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Cytochromeの電子の流れの順序



阻害剤と電子の流れ







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Figure 22-16 X-Ray structure of ferredoxin from Peptococcus aerogenes.



Figure 22-17a Oxidation states of the coenzymes of complex I. (a) FMN.



Figure 22-17b Oxidation states of the coenzymes of complex I. (b) CoQ.



最新の呼吸鎖構造図



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Figure 22-18 Electron microscopy–based threedimensional structures of Complex I.



The intricacies of complex II



Figure 22-19 X-Ray structure of *E. coli* quinol–fumarate reductase (QFR) in complex with its inhibitor oxaloacetic acid (OAA). (a) Ribbon diagram. (b) edge-to-edge distances.



Figure 22-20 Active site interactions in the proposed mechanism of the QFR-catalyzed reduction of fumarate to succinate.