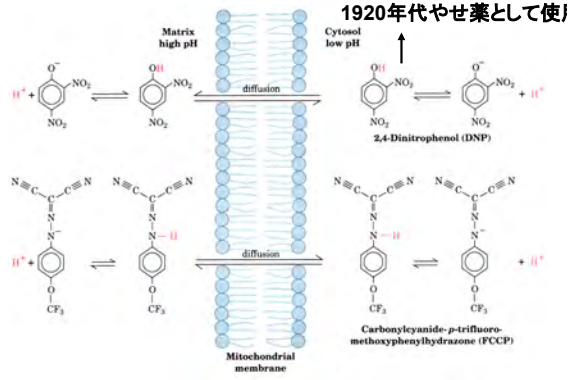
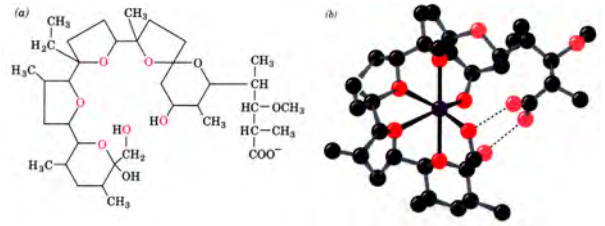


**プロトフォア(脱共役剤)**



**モネンジン**

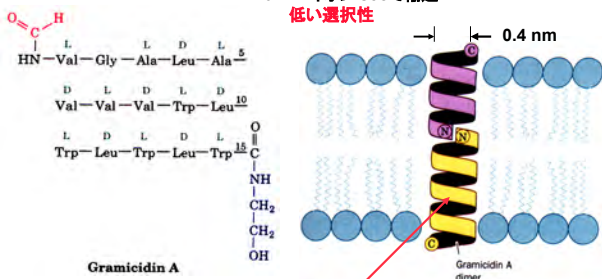
Na<sup>+</sup>と結合する線状ポリエーテルカルボン酸



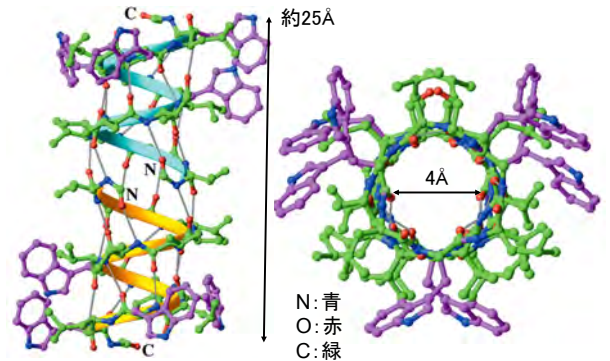
**グラミシジン**

15アミノ酸からなる直鎖ペプチド  
チャネルを作って一価のカチオンを通す

10<sup>7</sup> K<sup>+</sup> イオン/secで輸送  
低い選択性

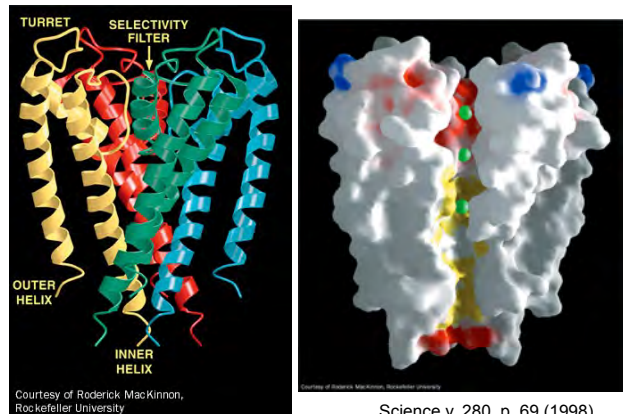
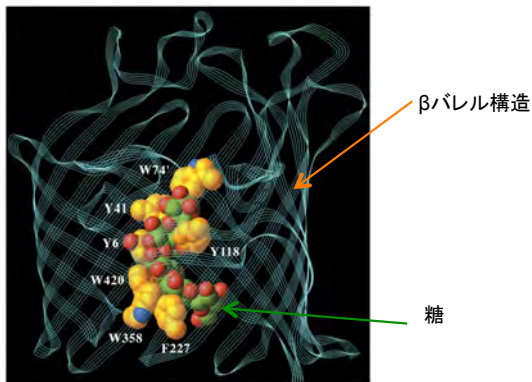


βヘリックス輸送  
低い選択性

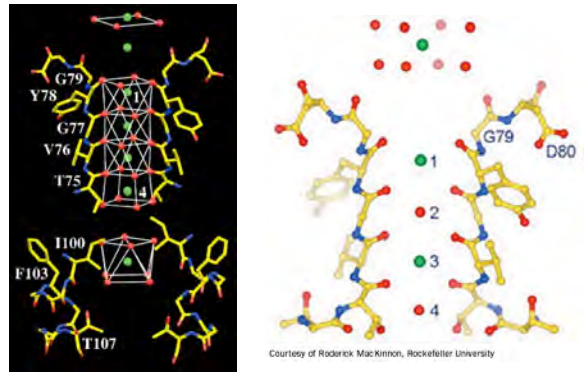
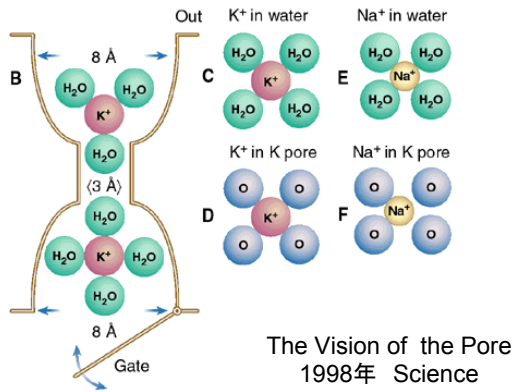


NMR structure of gramicidin A embedded in a dimyristoyl phosphatidylcholine bilayer.

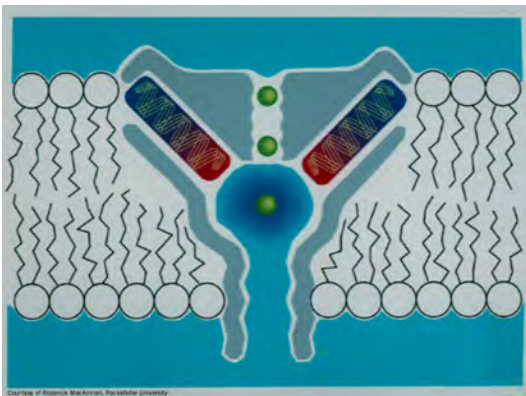
**マルトポリンの構造**



X-Ray structure of the KcsA K<sup>+</sup> channel from *Streptomyces lividans*.

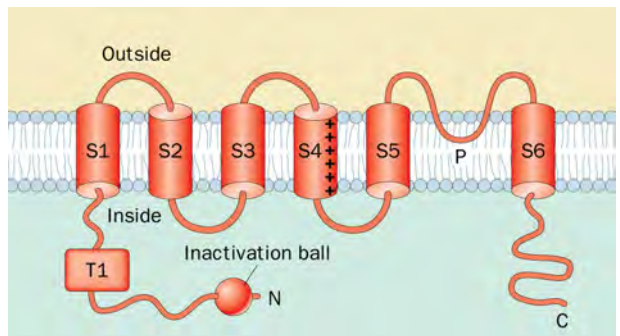


Portions of the KcsA K<sup>+</sup> channel responsible for its ion selectivity viewed similarly



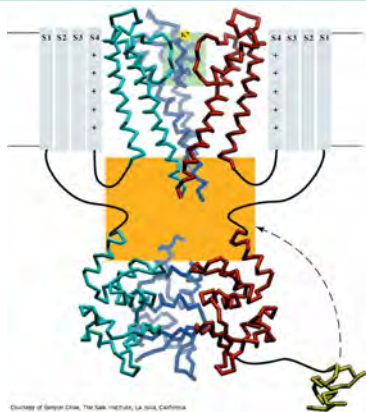
A schematic diagram of the KcsA K<sup>+</sup> channel

Predicted secondary structure and membrane orientation of voltage-gated K<sup>+</sup> channels.



Kvチャネル

Composite model of the KV channel.



電位依存KV channelのX線構造.

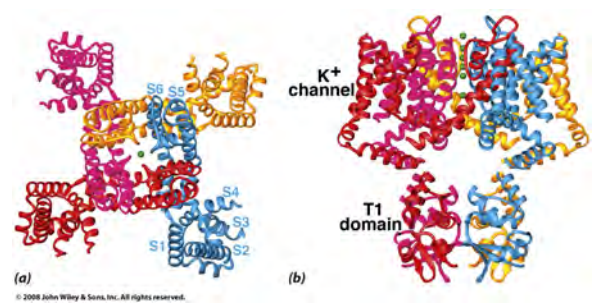
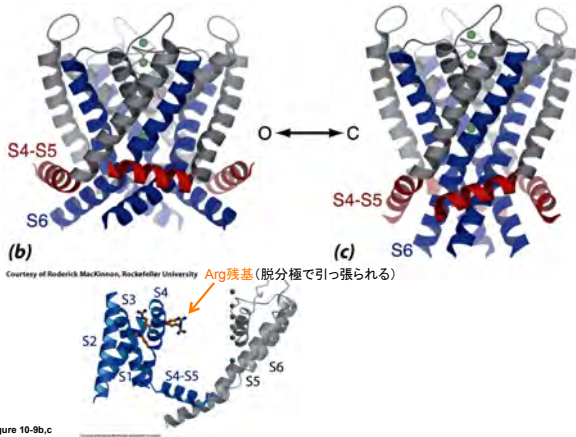


Figure 10-8

電位依存KV channelの膜貫通ドメインの動き.



Peter Agre



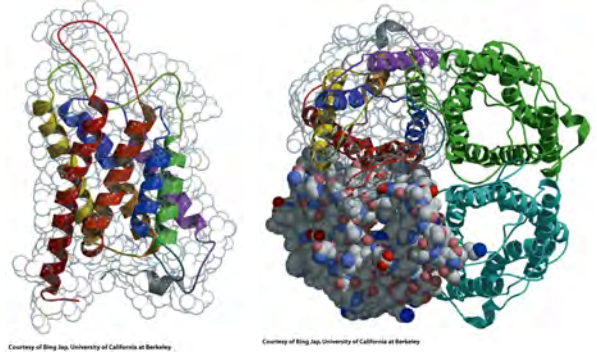
Roderick MacKinnon

The Nobel Prize in Chemistry 2003 was awarded for "for discoveries concerning channels in cell membranes" jointly with one half to Peter Agre "for the discovery of water channels" and with one half to Roderick MacKinnon "for structural and mechanistic studies of ion channels".

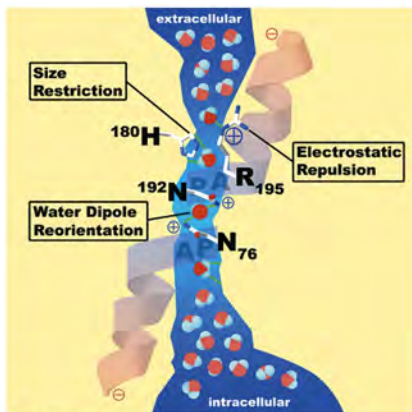
2013年1月 名古屋大学豊田講堂にて



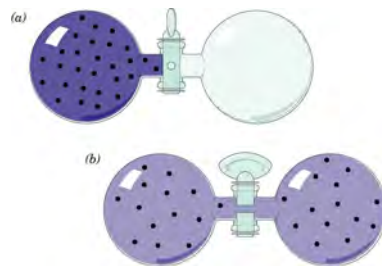
ウシ赤血球アクアポリンAQP1のX線構造



アクアポリンAQP1の水分子透過機構のモデル



Thermodynamic Principles



## 熱力学の法則

1) エンタルピーの定義:  $H = U + PV$

$$\Delta H = \Delta U + P\Delta V, \quad \Delta U = \Delta Q - \Delta W$$

(第一法則)

$$\Delta H = \Delta Q - \Delta W + P\Delta V = \Delta Q - \Delta W'$$

2) エントロピー:  $S \quad dS = dQ/T$  (可逆過程)

水の蒸発の  $\Delta H_{\text{vap}} = 40.7 \text{ kJmol}^{-1}$  で  
 $T = 373 \text{ °K}$  であるから

$$\Delta S_{\text{vap}} = 109.1 \text{ JK}^{-1}$$

蒸気になるときの  
エンタルピー変化

3) ギブスの自由エネルギー:  $G = H - TS$

$$\Delta G = \Delta H - T\Delta S \text{ (等温条件)}$$

$$1/2 \cdot mv^2 = 1/2 \cdot (2 \text{ kg}) \cdot (1 \text{ m} \cdot \text{s}^{-1})^2 = 1 \text{ Kg} \cdot \text{m}^2 \cdot \text{s}^{-2} = 1 \text{ Nm}$$

質量 2 kg が  $1 \text{ m} \cdot \text{s}^{-1}$  の速さで動いているものの  
運動エネルギーに 1J が対応

$$1 \text{ N} = 1 \text{ Kg} \cdot \text{m} \cdot \text{s}^{-2}$$
$$[\text{N} \cdot \text{m}] = [\text{J}]$$

$$0.24 \text{ cal} = 1 \text{ J} = 1 \text{ Kg} \cdot \text{m}^2 \cdot \text{s}^{-2}$$